

Complexes of n-diisopropoxythiophosphoryl-- '-phenylthiourea with a series of thiophilic metals

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Abstract

Potassium salts of N-diisopropoxythiophosphoryl-N'-phenylthiourea and diisopropoxyphosphorylthiobenzamide react with metal salts to form complexes $[\text{PhNHC(S)NP(S)(OPr-i)}_2]_2$ (M=Cd, Zn, Co, Hg, Pb) and $[\text{PhC(S)NP(O)(OPr-i)}_2]_2\text{Cd}$, whose structure was established by physicochemical methods. The complexes of Cu(II), Pb(II), and Hg(II) with the first ligand were found to be less stable than with the second. The high $4J_{\text{PH}}$ coupling constants (up to 8.0 Hz) for certain complexes is explained by the favorable configuration of the ligand and the presence of multiple bonds. ©2000 MAIK "Nauka/Interperiodica".
